

GALIBEV, V. I.; IVANOV, S. I.; YURZHENKO, I. A.

Activity of free radicals formed in the decomposition of
diacyl peroxides during styrene polymerization. Vysokom.
soed. 7 no. 1621746-17/32 O '65.

I. Odesskly gosudarstvennyy universitet.

(MIRA 1811)

CALIPIN, G.A., Cand Tech Sci -- (diss) "Selection of an
electrode wire for reconditioning auto tractor parts
^{f.2}
by electroplating ~~wire by direct spraying.~~ Len, 1958, 18 pp
(Min of Agr USSR. Len Agr Inst. Engineering Faculty) 120 copies
(KL, 23-58, 105)

- 54 -

GALIBIN, G.P.

Experimental study of the effect of consolidated synthetic resins
on an animal organism. Toks. nov. prom. khim. veshch. no.5:45-50
'63. (MIRA 17:9)

137-58-1-2145

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 1, p 292 (USSR)

AUTHORS: Yefremov, G. V., Galibin, V. A.

TITLE: On Thallium Colorimetry (K voprosu o kolorimetricheskem opredelenii talliya)

PERIODICAL: Uch. zap. LGU, 1957, Nr 211, pp 83-86

ABSTRACT: It is shown that determination of Tl by methyl red (I) requires decomposition of excess NaNO_2 which would decompose the I. As a result, the colored benzene layer becomes cloudy and this reflects upon the results of colorimetry. It is shown that at a Cl^- strength greater than 2 N urea does not reduce Tl because of the high stability of TlCl_4^- . To 4-5 cc of solution containing $\leq 50 \mu\text{g}$ Tl^+ , 2 cc 10 percent NaNO_2 and 5 cc concentrated HCl is added (the normality of the solution at the moment of oxidation should be ≥ 3). After 5 min the excess oxidizer is decomposed by 5 cc urea. The volume of solution is brought to 100 cc, 2 cc 0.2 percent solution I is added, and the Tl is extracted by two equal amounts of benzene (25 cc total). The acidity of the solution at the moment of extraction is 0.45-0.5 N. The benzene layer is then subjected to colorimetry. It is found that fuchsin,

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137-58-1-2145

On Thallium Colorimetry

parafuchsin, aniline, blue, basic blue, cetoglaucine, cetocyanine, and chrome green do not form colored compounds with $TlCl_4^-$. Methyl green produces a reversible reaction. Malachite green, basic bright green, and turquoise blue are suited to the colorimetry of Tl.

1. Thallium--Determination 2. Thallium--Colorimetric analysis 3. Colorimetry
--Applications

V. P.

Card 2/2

LEONOV A., V.A.; GALIBIN, V.A.

Association between zirconium and hafnium in zircons from pegmatites and gneisses from the White Sea region which enclose them. Dokl. AN SSSR 153 no.5:1168-1171 D '63.

(MIRA 17:1)

1. Leningradskiy gosudarstvennyy universitet im. A.A. Zhdanova.
Predstavleno akademikom N.V. Belovym.

LEONOV A., V.A.; GALEBIN, V.A.

Relation of zirconium to hafnium in the cyrtolites of the pegmatite veins of the Chupa region and in the zircon gneisses of the enclosing pegmatites. Min. i geokhim. no.1:163-180 '64. (MIRA 18:9)

ПЕЧАТНУ, В.Р.; ПРЕДОВСКИЙ, А.А.; РЕДОВСКИЙ, А.А.; ЧЕРНЫЙ, А.А.

Some characteristics of the distribution of impurity elements
in the biotites of crystalline schists and gneisses in the
northern part of the Lake Ladoga region. Vest. IGM 20 no.24:
5-14 '65. (MIRA 19:1)

1. Submitted May 15, 1965.

GALIBIN, YU. M.

C.A. V-48
Jan 10, 1954
Soils & Fertilizers

Methods of thorough improvement of sod-podzolized soils. P. M. Baley and Yu. M. Galibin. Izvest. Tiumenskogo Sel'skokhoz. Akad. (Magadan) No. 1(2), 18-28 (1953).—The plowed layer is deepened by having a section of the moldboard cut away so that the A_1 horizon is shattered and some of it mixed with the A_2 layer which is turned over and remains on the surface. No material from below is brought up to the surface, as is the case when deepening is done gradually by subsoiling. Another method of deepening the plowed layer is simply disrupting the soil to the desired depth without bringing up subsoil material. Expts. with these 3 methods show the highest improvement of the phys. condition of the soil, as well as its chem. properties, takes place when a section of the moldboard is cut out. This method gave more available P, more CO_2 release, and higher NO_3 content. The yield was also the highest. J. S. Joffe

(3)

GALIBIN, Yu. M.

"The Effectiveness of Ways of Deepening the Soils of the Southern Zone of the Moskovskaya Oblast";

dissertation for the degree of Candidate of Agricultural Sciences (awarded by the Timiryazev Agricultural Academy, 1962)

(Izvestiya Timiryazevskoy Sel'skokhozyaystvennoy Akademii, Moscow, No. 2, 1963, pp 232-236)

GALIBINA, A.I.

VORONOVA, A.M.; GALIBINA, A.I.

Treatment of cardiovascular insufficiency with intravenous erysimine.
Sov.med. 19 no.1:45-50 Ja '55. (MLRA 8:4)

1. Is fakul'tetskoy terapevticheskoy kliniki (dir. chlen-korrespondent Akademii meditsinskikh nauk SSSR prof. D.D.Yablokov) Tomskogo meditsinskogo instituta imeni V.M.Molotova (dir. prof. S.P.Khodkevich).

(CARDIAC GLYCOSIDES, therapeutic use,
Erysimum canescens glycoside erysimine)

YABLOKOV, D.D., prof.; GALIBINA, A.I., dotsent

Complications during antibacterial therapy of patients with
cavernous pulmonary tuberculosis. Sov. med. 24 no. 5:37-43
My '60. (MIRA 13:10)

1. Iz fakul'tetskoy terapevticheskoy kliniki (zav. kafedroy -
prof. D.D. Yablockov) Tomskogo meditsinskogo instituta (dir. -
prof. I.V. Toroptsev).
(TUBERCULOSIS)

GALIBINA, E.A., kandidat tekhnicheskikh nauk; DILAKTORSKIY, I.L., doktor geologo-mineralogicheskikh nauk

Investigation of the changes in the structure of the liquid phase of shale ash adhesive during the period of setting and initial hardening. Eesti tead akad tehn fuus 9 no.3:218-224 '60. (EEAI 10:3)

1. Institut stroitel'stva i stroitel'nykh materialov, Akademii nauk Estonskoy SSR.
(Adhesives) (Shale) (Ash)

"APPROVED FOR RELEASE: 07/16/2001

CIA-RDP86-00513R000614110007-1

GALIBINA, I.V.

Original and future orbits of long-period comets. Biul.Inst.teor.
astron. 9 no. 7:465-498 '64. (MIRA 17:9)

APPROVED FOR RELEASE: 07/16/2001

CIA-RDP86-00513R000614110007-1"

SOV/112-58-2-1863

Translation from: Referativnyy zhurnal, Elektrotehnika, 1958, Nr 2,
pp 10-11 (USSR)

AUTHOR: Vodop'yanov, K. A., and Galibina, G. I. (Dielektricheskiye
TITLE: High-Frequency Dielectric Losses in Crystals poteri v kristallakh
na vysokoy chastote)

PERIODICAL: Izv. Tomskogo politekhn. in-ta, 1956, Vol 91, pp 269-277

ABSTRACT: Experimental data are presented that characterize dielectric losses in simple and compound crystals with or without polar molecules. It is pointed out that a substantial difference in dielectric losses exists between crystals that have and do not have polar molecules. On the grounds of an experimentally determined relationship between $\tan \delta$ and crystal-lattice energy in alkali-halide crystals, a conclusion is drawn that $\tan \delta$ can be accepted as a basic characteristic of the substance and its structure, and that negligible impurities which do not affect the crystalline structure, play a secondary role. In the solid-solution systems of isomorphous NaCl - NaBr, KCl - KBr, and KI - KBr, crystals $\tan \delta$ vs. chemical composition is expressed as a curve with a maximum. The $\tan \delta$ maximum is more pronounced at lower frequencies and higher

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SOW/112-58-2-1863

High-Frequency Dielectric Losses in Crystals

temperatures. Mixed crystals have a higher $\tan \delta$ than that of pure alkali-halide crystals. On the basis of an investigation of temperature/frequency curves of $\tan \delta$ and ϵ in polar crystalline dielectrics (talcum, gypsum, blue vitriol), an essential role of crystallization water molecules is noted. Presence of temperature and frequency maxima of $\tan \delta$ with a corresponding variation of ϵ points to a relaxation mechanism of losses in polar crystals that is due to orientation of polar molecules or their radicals. Number of temperature-and-frequency maxima of $\tan \delta$ is determined by crystalline structure; their positions are determined by activation energy value of weakly bound dipoles. The value of a $\tan \delta$ maximum is determined by the number of dipoles taking part in the dielectric relaxation. Bibliography: 10 items. Sibirskiy fiziko-tehnich. in-t (Siberian Physics-and-Technology Institute), Tomsk.

K.A.V.

Card 2/2

AUTHORS: Vodop'yanov, K. A., Galibina, G. I. 48-22-3-15/30

TITLE: Dielectric Losses in Simple and Mixed Alkaline-Halogen Crystals at a High Frequency (Dielektricheskiye poteri v prostykh i smeshannykh shchelochno-galoidnykh kristallakh na vysokoy chastote)

PERIODICAL: Izvestiya Akademii Nauk SSSR, Seriya Fizicheskaya, 1958
Vol. 22, Nr 3, pp. 288-295 (USSR)

ABSTRACT: In the present paper the authors investigated the dielectric losses and the dielectric constant of a series of simple crystals, as well as of complex crystals. The loss angle changes with transition from one crystal to the other corresponding to the changes of a series of physico-chemical characteristic features of these crystals (table 1). The loss angle increase according to the increase of the distance between the ions and consequently with a reduction of the energy of the crystal and of its melting point. The changes of $\tan \delta$ are also subject to the rules governing the changes of physico-chemical characteristics. In order to understand the mechanism of the dielectric losses in the crystals, the change of the loss angle according to temperature and frequency must be known. The dependence of

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Dielectric Losses in Simple and Mixed Alkaline-Halogen Crystals 48-22-3-15/30
at a High Frequency

the frequency of the loss angle of the NaCl- and KCl -crystals with in the frequency range of $3 \cdot 10^5$ to 10^8 cycles per second as well as the temperature-dependence for all crystals in the temperature-interval of from 200 to 240°C were obtained. When measuring conductivity, the loss angle caused by the conductivity was obtained (table 2). The investigation results showed that, besides conductivity losses still other losses additionally occur in the observed crystals. Relaxation processes can develop in the alkaline-halogen crystals located in the electric field. The systems of solid solutions NaCl-NaBr, KCl-KBr KCl-RbCl, which are considered to be especially stable, were investigated. The X-ray analysis of the investigated crystals showed that most of them are really solid solutions. The measurements of the loss angle at room-temperature at different frequencies show the dependence of the amount of the loss angle on the composition of the solid solutions (figure 4). The conductivity of some solid solutions was measured! It was found in this connection that the loss angle calculated according to conductivity was much smaller at lower temperatures than the one measured immediately at these temperatures. The two values were almost equal at higher temperatures. The mechanism of the

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Dielectric Losses in Simple and Mixed Alkaline-Halogen Crystals at a High Frequency

48-22-3-15/30

losses in simple and complex crystals is equal. The increase of the loss angle in complex crystals and the more obvious dependence on temperature and frequency are correlated with characteristic features of the structure of these crystals. The same dependences were also observed in solid solutions of the substances of the ceramic type (Ref 12). It is known that small admixtures exercise a marked influence on the characteristic of the crystals. (Ref 13). The conductivity changes considerably under their influence (Ref 14). The change of dielectric losses with the introduction of admixtures was investigated mainly in the frequency band 10^1 to 10^5 cycles per second (Refs 15, 16). It was found that the small quantity of admixture which does not change the loss angle of the crystal at room temperature, causes a very considerable change of the loss angle at higher temperatures. As results from the given diagrams, the temperature dependences of the loss angles in the case of crystals with admixtures are analogous to those of the corresponding simple crystals in the case of crystals with admixtures. They are characterized, however, by a more rapid increase of the loss angle. The admixture of CaCl_2

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Dielectric Losses in Simple and Mixed Alkaline-Halogen Crystals at a High Frequency 48-22-3-15/30

exercises the greatest influence on the amount of the loss angle of the KCl crystals. The investigated admixtures exercise a similar effect also on the mechanical properties of the simple crystals: CaCl₂ and PbCl₂ exercise the greatest influence, whereas that of BaCl₂ and MgCl₂ (Ref 13) is smaller. There are 9 figures, 3 tables, and 17 references, 11 of which are Soviet.

ASSOCIATION: Sibirskiy fiziko-tehnicheskiy institut pri Tomskom gos. universitete im. V. V. Kuybysheva (Siberian Institute of Physics and Engineering at the State University imeni V. V. Kuybyshev)

AVAILABLE: Library of Congress

1. Alkaline--Halogen crystals--Dielectric properties

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S/048/60/024/02/05/009
B006/B014

AUTHORS: Vodop'yanov, K. A., Galibina, G. I.

TITLE: Investigation of Dielectric Losses in Solid Dielectrics at High Frequencies

PERIODICAL: Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1960, Vol. 24, No. 2, pp. 142 - 149

TEXT: The article under review was read at the Second All-Union Conference on the Physics of Dielectrics (Moscow, November 20 - 27, 1958). The authors give a detailed survey of the data obtained (and already published) by themselves and other authors concerning the frequency- and temperature-frequency dependence of the angle of dielectric loss in ion crystals of different chemical composition. The first section contains details of the relationship between the angle of dielectric loss of alkali halide crystals and their chemicophysical properties. The absolute value of $\tan \delta$ was found to be determined by these properties. This means that the angle of dielectric loss increases with decreasing energy of the crystal lattice, decreasing melting temperature and heat of formation, and increasing lattice constant. In the high-frequency range dielectric

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Investigation of Dielectric Losses in Solid Dielectrics S/048/60/024/02/05/009
at High Frequencies B006/B014

losses in ion crystals are mainly caused by weakly linked ions which have relatively low activation energies. The second section deals with the effect of heat treatment on the dielectric losses in alkali-halide crystals. The authors discuss their own results as well as those obtained by Yu. A. Sikorskiy, M. A. Krivov, B. M. Gokhberg, A. Venderovich, B. Lapkin, D. N. Popov, B. N. Matsonashvili, B. I. Vorozhtsov, et al. In the third section, the authors discuss the results of their own investigations concerning the effect of impurities on the dielectric losses in crystals and give some details of the frequency dependence of $\tan \delta$ (cf. Figs. 2-4). Finally, it is said that the temperature- and frequency dependence of $\tan \delta$ in crystals containing impurities is similar to the dependence of $\tan \delta$ in ordinary crystals. This indicates that in the case of ordinary and impurity crystals the same mechanism is responsible for dielectric losses. The increase of $\tan \delta$ in impurity crystals is related to the disintegration of the crystal lattice due to the impurities (activation energy and ion dissociation energy are reduced). Thus, the number of free and weakly linked ions is increased. Ya. N. Frenkel (Ref. 16) is mentioned in this connection. The fourth section deals with the dielectric losses in mixed crystals. Concerning the dielectric losses in solid solutions of alkali-halide crystals the authors state the following: 1) The mechanism of dielectric losses in crystals of solid

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Investigation of Dielectric Losses in Solid Dielectrics S/048/60/024/02/05/009
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solutions is analogous to the mechanism in ordinary crystals. The losses are composed of ohmic and relaxation losses, and result from considerable dislocations of free and weakly linked ions under the action of the electric field. 2) The dielectric losses in crystals of solid solutions are higher than the losses occurring in ordinary crystals, which fact is related to the structural particularities of solid solutions. 3) The rules governing the change of the physico-chemical characteristics of solid solutions differ from the rules governing the change of these characteristics of ordinary single crystals, which fact depends on the respective structure. G. F. Yarushevskaya participated in the studies of the effect of impurities on $\tan \delta$ and the crystal structure. There are 4 figures, 1 table, and 23 references, 17 of which are Soviet.

Card 3/3

GALIBINA, I.V.

Determining the initial orbit of the comet 1925 VII. Biul.Inst.teor.
astron. 5 no.7:412-419 '53. (MLRA 7:5)
(Comets—1925)

8
GALIBINA, I.V., Cand Phys-Math Sci--(diss) "Determination of initial and
future orbits of long-term ~~predicted~~ comets." Len, 1958. 9 pp
(Acad Sci USSR. Main Astronom Observatory), 100 copies (U,46-58,137)

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GALIBINA, I.V.

Determination of original and future orbits of some
long-period comets. Biul.Inst.teor.astron. 6 no.9:630-370
'58. (MIRA 13:3)
(Comets---Orbits)

GALIBINA, I.V.

Investigating original and future orbits of comets with a
near-unity eccentricity. Biul. Inst. teor. astron. 9 no.1:
46-81 '63. (MIRA 16:8)

GALIBINA, I.V.; BARTENEVA, O.N.

Final orbit for the comet Johnson 1950 I. Biul. Inst. teor. astron.
10 no.3:192-203 '65. (MIRA 18:8)

9.9822,
3.5110

S/139/62/000/004/005/018
E032/E514

AUTHORS: Galibina, L.I. and Zuyev, V.Ye.

TITLE: Absorption of long-wavelength radiation by water vapour and CO_2 bands along oblique directions in the atmosphere

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Fizika, no. 4, 1962, 69-74

TEXT: In a previous paper (Izv. vuzov SSSR, Fizika, no. 3, 1961) the second of the present authors showed that the semi-empirical formulae given by J. N. Howard, D. E. Burch and D. Williams (J. Opt. Soc. Am., 46, No. 3, 4, 1956) can be used to compute the absorption of long-wavelength radiation in a real atmosphere. However, in the case of oblique propagation these expressions cannot be used immediately because the total and partial pressures which enter into them are functions of altitude. It is, therefore, necessary to know the functional form of the altitude dependence of these pressures. It is now shown that the integral absorption of long-wavelength radiation in the water vapour and CO_2 bands can be computed using the Howard-Burch-

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Absorption of long-wavelength ... S/139/62/000/004/005/018
E032/E514

Williams expressions:

$$\int_{v_1}^{v_2} A_v dv = cw^{1/2}(P + p)^k; \quad (1)$$

$$\int_{v_1}^{v_2} A_v dv = C + D \log \omega + K \log (P + p), \quad (2)$$

with the concentration ω and total pressure P of water vapour given by

$$\omega = \frac{a_o}{2 \ln 10} (1 - 10^{-h/5}) \quad (7)$$

$$P = \frac{1}{h} \int_0^h P_o e^{-z/8} dz = \frac{8P_o}{h} (1 - e^{-h/8}). \quad (8)$$

In the case of CO_2 the vapour concentration is given by

$$\omega = 240(1 - e^{-h/8}) \quad (10)$$

and the total pressure by Eq.(8). These expressions hold for vertical propagation in the atmosphere and must be multiplied by $1/\cos \psi$ in the case of oblique incidence, where ψ is the zenith

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Absorption of long-wavelength ... S/139/62/000/004/005/018
E032/E514

angle. The authors have carried out numerical calculations for heights up to 8 km and six different zenith angles. Full numerical tables are given for the absorption functions computed from

$$A = \frac{\int_{v_1}^{v_2} A(v) dv}{v_2 - v_1} \quad (12)$$

for the following bands of CO_2 and H_2O vapour: (15, 5.2, 4.8, 4.3, 2.7, 2.0, 1.6, 1.4 μ) and (6.3, 3.2, 2.7, 1.87, 1.38, 1.1, 0.94, 3.7 μ) respectively. The CO_2 calculations refer to the following values of the zenith angle: 0, 15, 30, 45, 60 and 75°, while those for water vapour refer to the following absolute humidities: 0.5, 1.0, 2.0, 4.0, 8.0, 14.0 g/m³. There are 2 tables.

ASSOCIATION: Sibirskiy fiziko-tekhnicheskiy institut pri Tomskom gosuniversitete imeni V.V. Kuybysheva
(Siberian Physico-technical Institute of the Tomsk State University imeni V. V. Kuybyshev)

SUBMITTED: September 8, 1961

Card 3/3

AKHIEVA, Y.A., CHUDINA, V.A., PETELINA, O.N.; IOFFE, R.M.; TSYPLAKOVA, N.A.;
PARTIGUL'VA, R.Z.

affectiveness of compound health-resort treatment of residual
phenomena following infectious diseases of the central nervous
system. Sbor. nauch. rab. vrach. san.-kur. uchr. profsoiuzov
no.1:29-32 '64. (MIFPA 18:10)

2. Pyatigorskiy nevrologicheskiy sanatori "Mashuk" (glavnnyy vrach
R.Z.Partigulova, nauchnyy rukovoditel' prof.S.M.Petelin).

GALIBINA YE. A.

Setting of shale ashes. N. L. Diluktoroff and E. A. Galibina. Izvest. Akad. Nauk Estov. S.S.R. 4, No. 1, p. 3-21 (1933).³—The hardening of mixts. of shale ash and water is the result of the formation of a no. of hydrated minerals during the setting of the mixt. Calculation time and temp. affect the properties of the hydrated products. The mech. properties were poorest for the mixt. made with ash calcined for 2 hrs. at 110°, somewhat better immediately after setting of a mixt. calcined for 2 hrs. at 1000°, considerably better after curing for 88 days, and greatly improved after 8 months. Ash which was calcined for 1 min. at 800-1000° showed considerable strength after setting for 2 hrs., and reached a max. after 28 days. The hydration products were investigated by thermal differential analysis and were found to consist of products resulting from various physicochem. processes, such as the portland-cement process, the lime-silicate process, the sintering of lime, gypsum hydration, and slow formation of hydrated MgO. Because of the high-free-CaO content, cement setting can be accelerated by hydraulically active addns. (e.g. 20% of calcined diatomite), and also by thermal treatment during the setting (40 hrs. of steaming at 80° caused all free CaO to combine). W. M. Sumberg

GALIBINA, Ye. A. Cand Chem Sci -- (diss) "Processes of Mineral Formation Occurring During ~~the~~ Warming of the Inorganic Part of Kukersit and the Solidification of ~~Schistous~~^{slate} ~~rock~~ Zones." Len, 1957.

19 pp with graphs, 20 cm. (Min of Higher Education USSR, Len Order of Labor Red Banner ~~Engineering~~^{Technological} Inst im Lensoviet), 100 copies
(KL, 27-57, 105)

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GALIBINA, Ye. A. and N. L. DILAKTORSKIY

"On the Nature of Hardening Processes in Slate Cinders" p. 337

"Synthesis and Structure of Mineral Inclusions in Glass and Gneissic Heavy Metal Ossicles"

Transactions of the Fifth Conference on Experimental and Applied Mineralogy and Petrography, Trudy ... Moscow, Izd-vo AN SSSR, 1958, pl. 5p.

reprints of reports presented at conf held in Leningrad, 26-31 Mar 1956. The purpose of the conf. was to exchange information and coordinate the activities in the fields of experimental and applied mineralogy and petrography, and to stress the increasing complexity of practical problems

SOV/23-59-1-1/10

AUTHORS: Dilaktorskiy, N.L., Doctor of Geological-Mineralogical Sciences, Galibina, Ye.A., Candidate of Technical Sciences

TITLE: On the Solidifying of Slate-Ash Binding Substances During Autoclave Treatment (O tverdenii slantsezol'-nykh vyazhushchikh pri avtoklavnoy obrabotke)

PERIODICAL: Izvestiya Akademii nauk Estonskoy SSR, 1959, Nr 1, pp 3-12 (USSR)

ABSTRACT: This is an account of a study of the autoclave treatment of slate-ash binding substances made by the Institut stroitel'stva i stroitel'nykh materialov akademii nauk Estonskoy SSR (Institute of Construction and Construction Materials of the Academy of Sciences of the Estonian SSR) in 1956 - 57, on three types of powdered slate K₁₃, K₂₅ and K₂₆ in a laboratory autoclave, under pressures of 4, 8, 12 and 16 atm. Having been tested mechanically, the specimens were milled and subjected to thermographical, chemical, petrographical and X-ray

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SOV/23-59-1-1/10

On the Solidifying of Slate-Ash Binding Substances During Autoclave Treatment

examinations. The results of chemical examinations made by worker of ISSM of the AS Estonian SSR, L.N. Tamm, are shown in table 1. The study showed that the mechanical firmness of specimens of slate-ash paste is not as high as those made of a solution. The greatest mechanical firmness was shown by specimens that underwent a 20 hour-long hydro-thermal treatment at 140° C. The firmest specimens made of a solution were those treated for 12 hours at 175° C and for 4 hours at 190° C. A thermographic examination of the products of hydration of the paste showed the presence of high-basic hydrosilicates of " α " type, hydrates of " β " type and low-basic hydrosilicates of CSH(B) type. Specimens made of slate-ash solutions with an addition of fine milled sand showed increased mechanical firmness after autoclave treatment. With the increase in soaking temperature, the slate-ash paste showed changes in phase structure. There are 4 sets of graphs,

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SOV/23-59-1-1/10

On the Solidifying of Slate-Ash Binding Substances During Autoclave Treatment

2 sets of photos, 1 table and 9 references, 8 of which are Soviet, and 1 American.

ASSOCIATION: Institut stroitel'stva i stroitel'nykh materialov Akademii nauk Estonskoy SSR (Institute of Construction and Construction Materials of the Academy of Sciences of the Estonian SSR)

SUBMITTED: July 11, 1958

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S/023/60/000/003/004/012
C111/C222

AUTHORS: Galibina, Ye.A., Candidate of Technical Sciences, and
Dilaktorskiy, N.L., Doctor of Geological-Meneralogical Sciences

TITLE: Investigation of the Variation of the Combination of the Fluid
Phase During the Solidification and Primary Hardening of the
Binding Material of the Ash of Shale

PERIODICAL: Izvestiya Akademii nauk Estonskoy SSR. Seriya Tekhnicheskikh
i Fiziko-Matematicheskikh nauk, 1960, No.3, pp. 218-224.

TEXT: Pulverized ash of kukersit was changed to a mash by a normal
addition of water, it was stirred and then it was brought into a vapor
chamber. In certain distances tests were taken out of the vapor chamber.
Under pressure (2000 kg/cm^2) the fluid phase of the mash being in the
hardening process was squeezed out of the test. After a filtering of the
water the chemical combination of these tests were investigated. It was
stated: 1) Immediately after the addition of water to the ash there
appear solutions being strongly supersaturated with calcium oxide which
contain a great number of SO_4^{2-} -ions. 2) The concentration of Ca^{2+} , SO_4^{2-} and
 OH^- ions in the stone to be formed decreases during the hardening process,
since at the one hand there separate crystalline re-creations $\text{Ca}(\text{OH})_2$

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S/023/60/000/003/004/012

C111/C222

Investigation of the Variation of the
Combination of the Fluid Phase During the
Solidification and Primary Hardening of the
Binding Material of the Ash of Shale

and $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ and on the other hand in the first hours of the hardening there appears the combination $3\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot 3\text{CaSO}_4 \cdot 3\text{H}_2\text{O}$ difficult to dissolve. 3) The combination of ions of the fluid phase can be influenced essentially by a variation of the solution velocity of the combinations appearing in the ash. The solvability of the combinations with K^+ and Cl^- is diminished by additions of hydrophyl kind which raise the supersaturation of the solutions with the Ca^{++} and SO_4^{--} ions. The addition of

hot water (65°) diminishes the solvability of calcium oxide and anhydride and enlarges the solvability of combinations with K^+ and Cl^- . 4) the hydrothermic treatment of the investigated binding material for an elevated temperature (80°) diminishes the concentration of the Ca^{++} ions and enlarges the number of the K^+ and SO_4^{--} ions in the solutions in the interval 3 minutes - 4 hours (calculated from the beginning of the setting). Consequently by chemical additions a degree of solvability of the components appearing in the ash can be reached so that the

Card 2/3

Investigation of the Variation of the
Combination of the Fluid Phase During the
Solidification and Primary Hardening of the
Binding Material of the Ash of Shale

8/023/60/000/003/004/012
C111/C222

✓

forming of the skeleton structure during the setting of the binding material of shale ash can be take place without a change of volume. The authors mention L.Oyt. There are 2 figures, 2 tables and 2 Soviet references.

ASSOCIATION: Institut stroitel'stva i stroitel'nykh materialov Akademii nauk Estonской SSR (Institute of Construction Engineering and Building Material of the Academiy of Sciences of the Esthonian SSR)

SUBMITTED: February 4, 1960

Card 3/3

GALIBINA, Ye.A., kand.tekhn.nauk; DILAKTORSKIY, N.L., dcktor geol.-mineral.-nauk

Effect of the hydration rate of lime and anhydrite on volume changes and plastic strength of oil-shale ash stone. Izv AN Est SSR Ser fiz-mat i tekhn nauk no.4:267-277 '6 ..

1. Academy of Sciences of the Estonian S.S.R., Institute of Building and Building Materials.

GALIBINA, Ye., kand. tekhn. nauk; DILAKTORSKIY, N., doktor geol.-mineral. nauk; TAMM, L.

Effect of the composition of the liquid and solid phases on the extent of the voluminal changes and strength of shale-ash binding agents [with summary in English]. Izv. AN Est. SSR, Ser. fiz.-mat. i tekhn. nauk 12 no.1:91-99 '63. (MIRA 16:5)

1. Academy of Sciences of the Estonian S.S.R., Insitute of Building and Building Materials.
(Oil shales) (Cement clinkers)

DILAKTORSKIY, N.L., doktor geologo-mineralogicheskikh nauk; GALIBINA, Ye.A.,
kand.tekhn.nauk; KREMERMAN, T.B., inzh.

Phase composition of shale ash and its effect on the physical
and chemical processes under normal hardening conditions. Stroimmat.
10 no.4:31-33 Ap '64. (MIRA 17:5)

GALIBINA, Ye., kand. tekhn. nauk

Ionic composition of solutions formed under normal hardening
conditions of different fractions of dustlike oil-shale ash.
Izv. AN Est. SSR. Ser. fiz.-mat. i tekhn. nauk 13 no.3:254-259
'64. (MIRA 17:11)

1. Scientific Research Institute of Building of the State Building
Committee of the Council of Ministers of the Estonian SSR.

GALIBINA, Ye.; KREMERMAN, T.; DILAKTORSKIY, N.

Phase composition of different fractions of oil shale ashes and its effect on hardening processes. Izv. AN Est. SSR. Ser. fiz.-mat. i tekhn. nauk 14 no. 4:642-650 . '65 (MIRA 19:2)

1. Nauchno-issledovatel'skiy institut stroitel'stva Gosudarstvennogo komiteta Soveta Ministrov Estoneskoy SSR.

CALIC, A.

Preparation for firing with mine throwers by the use of
different instruments. p. 65. VOJNI GLASNIK. (Jugoslavenska
narodna armija) Beograd

Vol. 9, No. 6, June 1955

SOURCE: East European Accessions List, (EEAL), Library of
Congress, Vol. 4, No. 12, December 1955

SERAFIMOV, K.; GEORGIEV, K.; GALIC, B.

Case of giant ulcer of the stomach. Acta chir. Jugosl. 10
no.1:75-78 '63.

1. Hirurska klinika Medicinskog fakulteta u Skopju (Upravnik
prof. dr B. Dragojevic).
(STOMACH ULCER) (PEPTIC ULCER PERFORATION)
(GASTRECTOMY) (LIVER DISEASES) (PANCREAS)
(DISEASES)

S

Celtic, Roman

FILE I BOOK INFORMATION

ruc/3938

અનુભૂતિ

International Conference on Electrical Measurements and Instruments, Imperial College, London, August 6-12, 1959. 250 pp. £5.00. London, E.C. 1. Printed in Great Britain.

EDWARD We can't have him come in. He can't speak.

PURPOSE: This document will familiarize plant managers and technical personnel of Tupperware Industry.

COVERAGE: The collection contains papers presented during the annual structural engineering symposium which took place in Taizhou (Ningbo) in 1999. The following problems were discussed:

on national orders, segment, and the extension of processes, application of rules in certain varieties, mechanical and economic. In addition to the

In this collection, the following papers were presented at the conference:
Leland in this collection, the following papers were presented at the conference:
Professor Dr. Jacob Leland, Electrical Engineering Department, Indiana
University, Bloomington, Indiana, and the following were presented by Professor

THE WORLD WAR: THIS SECTION IS A PICTURE OF FRENCH ARMED FORCES ON THE BATTLEFIELD, AND IT SHOWS HOW THEY ARE BEING USED IN THE FIELD OF WAR.

Division of Electrical Engineering, Zürich University, Winterthurerstrasse 190, 8401 Zürich, Switzerland; Robert H. Dicke, Institute for Space Sciences, Max-Planck-Institut für Physik und Kosmologie, Postfach 152, D-8046 Garching bei München, Germany

Proceedings of the International Conference on Scientific and Technical Cooperation in Using Atomic Energy for Scientific and Industrial Purposes, and a **Publication**, **Technical Manual for the Scientific and Industrial Purposes**, and a **Publication**, **Current Status and Prospects of the Field Nuclear Power Plants**, **Japan**. **Development and Prospects of the**

१०८ विष्णु वाचना विष्णु वाचना विष्णु वाचना

Walter Doctor, Doctor, Department of Electrical Engineering, Research Institute for Application of Industrial

Problem 9 The author describes the principles of measuring devices used in biological universities. There are 7 references: 6 German and

WILLIAM MILLARY,
BOSTON, MASS., made Boston Flats, Bridge Lane. WILLIAM
OLIVER, Bridge Lane, Boston, Mass.

The author describes characteristic features of radar equipment from World War II to the present. There are 3 references: 2 English and

W. H. Gaskins, 1820-1821, and James M. Clegg, 1821-1822.

— **James** **Platt,** *Delta Louie*, **describes** **single-lead** **and** **expander** **systems** **for** **radio** **transmitters** **using** **modulation** **resistors**, *radio* **frequency** **oscillators** **and** **references.** 6 **English** **and**

Frances' experience was similar to that of the other women in the department, though this

Review Article. In addition, Frequency domain, wavelet analysis, and other descriptive principles of inductive testing, its applications, and necessary algorithms.

Industrial Electro Metal Plant, Regent Application of Ultrasonic in Purifying Base Metals

The abstracts presented here are all methods for calculating hard metals. Some examples of the processes given. There are 4 references: 2 German, 1 Soviet, and 1 English.

Elect., Inv., Engineer, Vlado Njegos Plančić, Zadar. Plastics utilized in Electronics

The author describes the characteristic features of plastics used in aircraft types and characteristics. There are 14 references: 3 English and 1 German.

**Rudolf Hertz, Berliner. Application of Electrode Automatic Control
in Industrial Welding Processes.**

The author describes principles and applications of the various processes. There are 9 references: 4 English, 3 German, and 2 French.

卷之三

APPROVED FOR RELEASE: 07/16/2001

CIA-RDP86-00513R000614110007-1"

"APPROVED FOR RELEASE: 07/16/2001

CIA-RDP86-00513R000614110007-1

GALIC, Roman, inz. (Zagreb)

New curves of propagations for the frequencies 40 - 960 MHz.
Elektrotehnika Hrv 4 no.1/2:3-4 '62.

APPROVED FOR RELEASE: 07/16/2001

CIA-RDP86-00513R000614110007-1"

JELIC, R.; PEROVIC, S.; GALIC, R.

Esophago-bronchial fistula. Tuberkuloza l7 no.3:268-276
My-Je '65.

I. Medicinski centar, Zadar, odjel za rentgenologiju, odjel
za djecje bolesti i odjel za zarazne bolesti.

GALIC, Wacław, mgr inż.; WASOWSKI, Janusz, mgr inż.

The shipyard in Danzig in the light of certain statistical data. Bud. okretowe Warszawa 8 no.11:398-405 N°63.

1. Stocznia Gdanska, Gdańsk.

WOJEWSKI, Alfons; GALICA, Eleonora

Neurofibromatosis of the urinary bladder. Polski przegl. chir. 33
no.1:75-78 '61.

1. Z Oddzialu Urologicznego P.A.M. w Szczecinie. Kierownik: zast.
prof. dr A. Wojewski.

(NEUROFIBROMATOSIS case reports)
(BLADDER neopl)

HILCZER, Bozena; GALICA, Jerzy

Effect of gamma ^{60}Co radiation on the temperature hysteresis of triglycine sulfate. Prace matem przyrod Poznan 11 no. 2:199-206 '64.

1. Department of Dielectrics, Institute of Physics, Polish Academy of Sciences, and Department of Experimental Physics, A. Michiewicz University, Poznan.

PIEKARA, A.; STANKOWSKI, J.; SMOLINSKA,S.; CALICA, J.

The ammonia maser of the Poznan Center. Postepy fizyki 15
no.5:565-568 '64.

1. Department of Dielectrics, Institute of Physics, Polish
Academy of Sciences, Poznan, and Department of Experimental
Physics, A. Mickiewicz University, Poznan.

CALICEK, Stanislav, inz.; JIRA, Lubor, inz.

Use of modern calculation methods in designing broad gauge railroad tracks. Doprava no.4:307-308 '64.

Graphomat, unspecified, exp.

Graphomat Zuse Z 64, an automatic rectangular plot climate recorder. Inz. naivky 15 no. 1; 66-01-3 '65.

1. Calculation Laboratory of Transportation, Prague.

GALICH, I. I.

Ship communication and navigation devices; textbook Leningrad Gosl izd-vo
sudostroit. lit-ry, 1952. (Mic 55-3193)

Collation of the original, as determined from the film: 340 p.

Microfilm Slavic 347 AG

GALICH, Iliodor Illarionovich; GEDIKE, S.R., otv.red.; KRUGOVA, Ye.A.,
red.; TSAL, R.K., tekhn.red.

[Electric and radio navigation equipment] Elektro- i radio-
navigatsionnye pribory. Leningrad, Gos.soiuznoe izd-vo sudo-
stroit.promyshl., 1959. 198 p. (MIRA 13:2)
(Aids to navigation) (Nautical instruments)

GALICH, Iliodor Illarionovich, KITAYENKO, G.I., retsenzent; TIMOFEYEV,
E.S., retsenzent; BOYTSOV, A.Ye., retsenzent; NIKITINA, M.I.,
red.; TSAL, R.K., tekhn. red.

[Electric control systems of ships] Sudovye elektricheskie
ustanovki upravleniya. Leningrad, Sudpromgiz, 1962. 259 p.
(MIRA 16:2)

(Ships--Electric equipment)
(Ships--Electronic equipment)

TSIPEROVICH, O.S. [Tsypetrovych, O.S.]; GALYCH, I.P. [Halych, I.P.]

Chromatographic investigation of egg and serum albumins and
chymotrypsinogen in columns with hydroxylapatite. Ukr.biokhim.
zhur. 34 no.5:666-677 '62. (MIRA 16:4)

1. Institut biokhimii AN UkrSSR, Kiyev.
(ALBUMIN) (CHYMOTRYPSINOGEN)
(CHROMATOGRAPHIC ANALYSIS)

TSYPEROVICH, A.S. [TSyperovich, O.S.]; GALICH, I.P. [Galich, I.P.]

Fractionation of bovine gamma globulin on columns of carboxy-
methyl- and diethylaminethyl cellulose. Ukr. biokhim. zhur.
35 no.6:931-941 '63. (MERA 18:7)

1. Institut biokhimii AN UkrSSR, Kiyev.

GALICH, I. V.

Fuel Abst:

Vol. 15 No. 4

Apr. 1954

Natural Solid Fuels:

Winning

(2)

✓ 2716. SQUEEZING OUT OF COAL DURING WORKING OF BEAM L3 IN MINE 17/18
(STALINUGOL COMBINE). Galich, I.V. (Ugol (Coal), Oct. 1953, 16-19). The
phenomenon and the conditions for its occurrence are described. (L).

GALICH, I.V.

Displacement of a coal layer in connection with the removal of
coal. Trudy VNIMI no.29:121-127 '54. (MLRA 8:3)
(Coal mines and mining)

ALICH, I.V., inzhener-geolog

Mining under bore-hole areas. Ugol' Ukr. 5 no.10:34-36 O '61.
(MIRA 14:12)
(Coal mines and mining)

GALICH, I.V., inzhener-geolog

Special method of reinforcing the ends of underground drain
bore holes. Ugol' 36 no.10:35-36 o '61. (MIRA 14:12)

1. Shakhta No.1 "TSentral'naya" tresta Krasnoarmeyskugol'.
(Mine drainage)

GALICH, I.V.

Rock cleavage and the deflection of the shafts of boreholes.
Razved. i okh. nedr 28 no.8:51-53 Ag '62. (MIRA 15:3)

1. Krasnoarmeyskiy trest ugol'nykh predpriyatiy Donbassa.
(Donets Basin--Boring)

GALICH, I. V., inzh.-geolog

Special means of sinking shafts in the western Donets Basin.
Ugol' Ukr. '7 no.4:36-37 Ap '63. (MIRA 16:4)

(Donets Basin—Shaft sinking)

GALICH, I.V., inzh.-geolog (g. Novosokonomiceskoye)

Rock cleavage and manifestation of rock pressure. Ugol' 40
no.4:37-39 Ap '65. (MERA 18:5)

TSYPEROVICH, A.S. [TSyperovych, O.S.]; GALICH, I.P. [Halych, I.P.]

Crystallization of α -amylase of *Aspergillus oryzae* from Soviet
enzyme preparations. Ukr. biokhim. zhur. 37 no.1:14-23 '65.
(MIRA 18:5)

1. Institut of Biochemistry of the Academy of Sciences of the
Ukrainian S.S.R., Kiyev.

English, limited (Grad. Stan.)

Dissertation: "Tar Production from the hydrocarbons of Diesel fuels by oxidizing in the liquid Phase." Cand Chem Sci, Institute of Petroleum, Acad Sci USSR, 17 Jun 54. (Vechernyaya moskva, Moscow 8 Jun 54)

SO: Sov 313, 23 Dec 1954

GALICH, P. N.

b) Liquid-phase oxidation of α -undecene, α -heptylbenzene, n -heptylcyclohexane, and 1-methyldecahydroanthracene by air oxygen. S. K. Sarieenko and P. N. Galich. Akademijskie Gidrokarbony i Oksidantsy. Akad. Nauk Kazakh. SSR, Trudy Inst. 1958, 153-64. -- In liquid phase, air oxidation of hydrocarbons at 15° for 10-100 hrs, the 1st stage is the formation of peroxides which reaches a max. in 10 hrs. and declines to nearly 0 by 100 hrs. The kinetic curves of time vs. acid no., bromine no., sapon. no., peroxide content and O content are shown for n -undecene, n -heptylbenzene, n -heptylcyclohexane, and 1-methyldecahydroanthracene. The tendency to form tar during oxidation declines in order: undecene, 1-methyldecahydroanthracene, heptylcyclohexane, heptylbenzene. Tar formation begins to be noticeable after some 60 hrs. The gradual decline of the rate of oxidation and its practical cessation in the presence of as yet unreacted materials is ascribed to the neg. catalytic activity of the tars which act as antioxidants.

G. M. Kosolapoff

GALICH, P.N.

3504. TAR FORMATION DURING OXIDATION OF HYDROCARBONS
 (HYDROCARBON) GROUP COMPOSITION OF DIESEL FUELS ON THE DIRECT

Sergienko, S.R. and Galich, P.N. (Zh. prikl. khim. (J. appl. Chem., Moscow), 1956, vol. 29, 1568-1577; USSR, in Chem. Abstr., 1957, vol. 51, 6166). The was removed from a 300-3650 fraction of diesel fuel (22) by distillation on silica gel and was separated, respectively, into fractions of aromatic hydrocarbons (I), 20.7% and paraffins-cycloparaffins. The latter fraction was treated with urea and separated into n-paraffins (II), 32.4% and isoparaffins-cycloparaffins (III), 62.1%, leaving a residue (IV), 4.8%. Each fraction was oxidized by air at 150° in 2 parallel furnaces. The properties of the products were plotted as functions of the oxidation time. The properties of I did not change appreciably, a slight darkening indicated a slow condensation. Those of IV changed a little more. The plots of II and III gave similar functions. The proportion of peroxides rose sharply to a maximum within 2 hours for III and 9 hours for II. Beyond this maximum the acid number, the saponification number, and the proportion of oxygen-containing substances increased continuously. Tar formation became noticeable in 50 hours with III and in 75 hours with II. Oxidation of diesel fuel is retarded not so much by I as by the products of cleavage of the peroxides formed.

11. EFFECT OF

OXIDATION ON CHIDATION

12. C.H.

13. MT

GALICH, P. M.

Tar formation in the oxidation of hydrocarbons. III.
Effect of aromatic hydrocarbons on the course of the oxidation
of hydrocarbon mixtures. S. K. Serbinenko,
Galich, and V. I. Leytev. Zhur. Prakt. Khim. 20, 1746-1750
(1950); cf. C.A. 45, 7406. In air oxidation of the hydrocarbons from the kerosene-cycloparaffin fraction of Diesel
fuel, the initial stage involves the formation of peroxides

which reaches a max. in about 2 hrs. after which the peroxide content decreases steadily. Peroxide formation precedes the formation of acidic, hydrolyzable, and Off compds. Addn. of 10% or less of diisopropylbenzene to the charge does not affect the oxidation kinetics, but 30% or more almost completely stops oxidation. The retarding effect is not produced by diisopropylbenzene itself, but by the products of decomp. of its peroxides. Max retardation is caused by addn. of 20-6% aromatic compds. G. M. K.

Distr: 4E3d/4E4f

*Causes of deposit formation in Diesel oil during storage
S. R. Serdenko and P. N. Galich, Zhur. Promst. Khim., No. 20,
1953-54 (1957).--Aromatic hydrocarbons and cyclo-
and isoparaffin fractions obtained from Diesel oil by chro-
matographic methods (cf. C.A. 50, 8186) as well as Diesel oil
were kept 2-5 years in open glass vessels at room temper-

The aromatic hydrocarbons were most susceptible to de-
posit formation; after 2 years 0.2% of asphaltene had
formed. The cyclo- and isoparaffins showed an extended
induction period so that after 2 years deposits were absent
but a high concn. of peroxides was found. Deposit forma-
tion was at first rapid, reached a max., and decreased subse-
quently, apparently retarded by oxidation products.

I. Bezugovits

9/16

GALICH, P.N.

SERGIYENKO, S.R.; IZMAYLOV, N.A.; SPIVAK, L.L.; GALICH, P.N.

Potentiometric methods of investigation of high-molecular weight
compounds in petroleum. Zhur.anal.khim.10 no.5:315-322 S-0 '55.
(MLRA 9:1)

I.Institut nefti AN SSSR, Moskva i Khar'kovskiy gosudarstvennyy
universitet imeni Ger'kego.
(Potentiometric analysis) (Acidity) (Petroleum)

AID P - 3572

Subject : USSR/Chemistry
Card 1/1 Pub. 152 - 9/20
Authors : Sergiyenko, S. R. and N. P. Galich
Title : Tar formation during the oxidation of hydrocarbons
Periodical : Zhur. prikl. khim., 28, 7, 735-744, 1955
Abstract : The mechanism of oxidation of C₁₁-C₁₃ hydrocarbons at 150°C has been studied. The oxidation products of n-heptyl benzene contained 13% tar and 27% asphaltenes, and those of n-undecene-1, 85% tar and no asphaltenes. Four tables, one diagram, 8 references, 4 Russian (1937-1955).
Institution : None
Submitted : D 25, 1954

GALICH, P. N.

14034. Potentiometric methods of studying multi-molecular compounds of petroleum. II. Potentiometric determination of saponification values in solutions of petroleum resins. S. R. Sartakov, P. N. Galich, N. A. Lamalov, and L. L. Splyut' (Inst. of Petroleum Acad. Sci. USSR, Moscow). Zhur. Anal. Khim., 1956, 11 (6), 731-734. The sample is dissolved in a mixture of ethanol and benzene (2:5), then mixed with a 5 to 10-fold excess of ethanolic 0.1 N KOH and heated on a water bath under a reflux condenser for 1 hr without admission of CO_2 . The hot soln. is titrated potentiometrically with 0.1 N HCl in a CO_2 -free atmosphere, glass and calomel electrodes being used.

G. S. Sartakov

14034
14034

CALICH, P.N.

✓ Potentiometric method for the study of free radical petroleum compounds. III. Potentiometric determination of peroxide numbers. S. R. Semenko, P. N. Calich, and L. L. Sivak. (Petrochemistry, 1971, 12, 1251). *J. Am. Chem. Soc.*, 1971, 93, 1171 (1971). *J. Org. Chem.*, 1971, 36, 9185. Dissolve a weighed sample in a mixt. of abs. iso-PrOH 20 and glacial AcOH, ml. Add 10 ml. satd. NaI soln. in abs. iso-PrOH. Stir for 5 min., add 5 ml. NaO₂ and titrate potentiometrically with 0.1*N* Cu²⁺ while passing N through the soln. Specific examples are given. M. Heschl

GALICH, P.N.

SERGIYENKO, S.R.; DAVYDOV, B.E.; GALICH, P.N.

Molecular and surface properties of petroleum tars. Part 7.
Article No.7. (MIRA 11:4)
(Petroleum) (Surface chemistry)

GALICH, P.N.

- 5(3) 12(4) PAPER 1 KOK EXPEDIMENTA - 607/1222
- Academy and USSR Institute report
Study, v. 12 (Transactions of the Petroleum Institute, USSR, Academy of Sciences, Vol. 12) Moscow, Izd-to AN SSSR, 1958. 395 p. Metal slip inserted. 1,700 copies printed.
- M. N. Sereyenko, Professor; Ed. of Publishing House: K. G. Klyashevsky Tech. Ed.; V. V. Golubeva.
- PURPOSE:** The book is intended for scientists, engineers, and technicians in the petroleum industry.
- CONTENTS:** This collection of articles describes the results of studies on the chemistry and technology of petroleum and gas conducted in the Laboratory of the Petroleum Institute, Academy of Sciences, USSR, in 1956 and 1957. A new section "Research Methods and Technology of Petroleum" has been added. In the collection of articles, a list of publications published by the scientists of the Institute in 1956 and 1957 and a list of dissertations by the associates of the Doctor's and Candidate's degrees presented in 1956 and 1957 at open sessions of the Academic Councils of the Petroleum Institute, Academy of Sciences, USSR, are given.
- E. N. Zabikheva, P. V. Korovayev, I. A. Masyrov, and V. V. Shchekin. Change in the Activity of Silica Gel in the Chromatographic Separation of Hydrocarbons
- Sel'zhenko, G. D., M. M. Rusakov, Ye. S. Potokritova, and N. A. Shchukina. Study of the Absorption Spectra of Benzene Crudebaryn- and Cyclohexyl-Benzene Derivatives in the Near Ultraviolet Region
- Card 2/9
- Sereyenko, S. R., Ye. Ya. Sereyenko, and B. T. Gordach. Investigation of the Composition and Properties of High-Molecular-Weight Hydrocarbons and Part of Oily-Gum Petroleum
- Sereyenko, S. R., B. T. Gordach, A. D. Il'manovich, and V. A. Shchukin. Some Physicochemical Properties of Petroleum Asphaltenes and Tar Substances. Part 11.
- Sereyenko, S. R., and B. T. Gordach. Composition and Properties of the Gas Fraction of Radishkino Petroleum. Part 13
- Sereyenko, S. R., and B. T. Gordach. Iso-Structure Transformations of High-Molecular-Weight Aromatic Hydrocarbons of Radishkino Petroleum. Part 15
- Sereyenko, S. R., Ye. Ya. Lebedeva. Chemical Nature of Saturated High-Molecular-Weight Hydrocarbons of Radishkino (Derodino) Petroleum. Part 17
- Sereyenko, S. R., Ye. Ya. Lebedeva, and Ye. V. Labedeva. Chemical Nature of Saturated Light-Molecular-Weight Hydrocarbons of Radishkino (Derodino) Petroleum. Part 19
- Sereyenko, S. R., and A. A. Mikhnevskaya. The Chemical Nature of High-Molecular-Weight Mono- and Dicyclic Aromatic Compounds of Radishkino (Derodino) Petroleum. Part 20
- Sereyenko, S. R., Ye. Ya. Lebedeva, and Ye. V. Radishkina. Hydrogenation of High-Molecular-Weight Mono- and Dicyclic Aromatic Compounds of Radishkino Petroleum in the Presence of a $\text{W}_2\text{-NiB}$ Catalyst under 1114°C Conditions. Paper 21
- Sereyenko, S. R., Ye. Ya. Lebedeva, and Ye. V. Radishkina. Isoprenization of Some Isolated from Radishkino Petroleum. Paper 22
- Sereyenko, S. R., Ye. Ya. Lebedeva, and Ye. V. Radishkina. Effect of the Depth of Distillation on the Composition and Properties of Heavy Radishkino Petroleum. Paper 23
- Sereyenko, S. R., Ye. Ya. Lebedeva, P. N. Galich, L. I. Butman, B. P. Savchenko, and M. I. Kravchenko. Effect of the Nature of the Raw Material and Oxidation Time on the Composition and Properties of Crude Oil Bitumens. Article 24

SERGIYENKO, S.R.; KORCHAGINA, V.I.; GALICH, P.N.; RUTMAN, L.I.; DAVYDOV, B.B.;
KRASAVCHENKO, M.I.

Effect of the depth of sampling on the composition and properties of
heavy residual stock. Article No.23. Trudy Inst.nefti 12:175-186
'58. (MIRA 12:3)

(Petroleum products--Analysis)

SERGIYENKO, S.R.; KORCHAGINA, V.I.; GALICH, P.N.; RUTMAN, L.I.; DAVYDOV, B.E.;
KRASAVCHENKO, M.I.

Effect of the nature of feed stock and the duration of oxidation on
the composition and properties of oxidized bitumens. Article No.24.
Trudy Inst.nefti 12:187-199 '58. (MIRA 12:3)
(Bitumen) (Petroleum--Refining)

5(3)

SCOV/80-32-3-31/43

AUTHORS: Sergiyenko, S.R., Semyachko, R.Ya., Galich, P.N.

TITLE: The Liquid-Phase Oxidation of High-Molecular Hydrocarbons of Petroleum (Zhidkofaznoye okisleniye vysokomolekulyarnykh ugle-vodorodov nefti)

PERIODICAL: Zhurnal prikladnoy khimii, 1959, Vol XXXII, Nr 3, pp 641-649 (USSR)

ABSTRACT: In high-molecular hydrocarbons of petroleum, compounds are contained which have condensed aromatic structures in their molecules. These compounds are the sources for the formation of resinous asphaltene substances. A genetic system exists in these hydrocarbons which may be represented by the following series: condensed bicyclic aromatic compounds → condensed polycyclic aromatic compounds → resins → asphaltenes. At an oxidation temperature of 150 - 175°C the asphaltenes prevail in the oxidation products. The paraffin-cycloparaffin hybrid compounds are transformed during oxidation in the liquid phase at a temperature of 150 - 175°C to peroxide compounds which in turn are transformed to acid saponifiable hydroxyl-containing oxygen compounds. The bicyclic aromatic condensed hydrocarbons con-

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The Liquid-Phase Oxidation of High-Molecular Hydrocarbons of Petroleum

densate most easily of all hydrocarbons, followed by the monocyclic aromatic compounds. The paraffin-cycloparaffin hydrocarbons oxidize more easily at 150°C than at 175°C.

There are 3 graphs, 3 tables, 1 diagram and 10 Soviet references.

SUBMITTED: May 16, 1957

Card 2/2

GALICH, P.N.; GOLUBCHENKO, I.T.; GUTYRYA, A.A.; GUTYRYA, V.S.; DOLINSKAYA, E.S.; MOZDOR, Ye.V.; NEYMARK, I.Ye.

Nature of cokelike deposits formed on CaC-type molecular sieves in the cracking of n. alkanes. Neftekhimiia 2 no.2:193-195 Mr-Ap '62.
(MIRA 15:6)

1. Institut khimii polimerov i monomerov AN USSR i Institut fizicheskoy khimii imeni Pisarzhevskogo AN USSR, Kiyev.
(Paraffins) (Cracking process)

GALICH, P.N.; GUTYRYA, A.A.; GUTYRYA, A.A.; GUTYRYA, V.S.; NEYMARK, I.Ye.

Certain features of the catalysis of alkanes over zeolites
(molecular sieves). Dokl.AN SSSR 144 no.1:147-150 My '62.
(MIRA 15:5)

1. Institut khimii polimerov i monomerov AN USSR i Institut
fizicheskoy khimii AN USSR. 2. Chlen-korrespondent AN SSSR (for
Gutryya).

(Paraffins) (Catalysis) (Zeolites)

GALICH PN

128

PHASE I BOOK EXPLOITATION

SOV/6246

Soveshchaniye po tseolitam. 1st, Leningrad, 1961.

Sinteticheskiye tseolity; polucheniye, issledovaniye i primeneniye
(Synthetic Zeolites: Production, Investigation, and Use). Mos-
cow, Izd-vo AN SSSR, 1962. 286 p. (Series: Its: Doklady)
Errata slip inserted. 2500 copies printed.

Sponsoring Agency: Akademiya nauk SSSR. Otdeleniye khimicheskikh
nauk. Komisiya po tseolitam.

Resp. Eds.: M. M. Dubinin, Academician and V. V. Serpinskiy, Doctor
of Chemical Sciences; Ed.: Ye. G. Zhukovskaya; Tech. Ed.: S. P.
Golub'.

PURPOSE: This book is intended for scientists and engineers engaged
in the production of synthetic zeolites (molecular sieves), and
for chemists in general.

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Synthetic Zeolites: (Cont.)

SOV/6246

COVERAGE: The book is a collection of reports presented at the First Conference on Zeolites, held in Leningrad 16 through 19 March 1961 at the Leningrad Technological Institute imeni Lensoveta, and is purportedly the first monograph on this subject. The reports are grouped into 3 subject areas: 1) theoretical problems of adsorption on various types of zeolites and methods for their investigation, 2) the production of zeolites, and 3) application of zeolites. No personalities are mentioned. References follow individual articles.

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Synthetic Zeolites: (Cont.)

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- Pavlova, S. N., Z. V. Driatskaya, and M. A. Mkhchiyan.
Application of Synthetic Zeolites in Determining the
Content of Normal Alkanes in Gasoline Fractions 253
- Galich, P. N., I. T. Golubchenko, A. A. Gutyrya, V. S.
Gutyrya, and I. Ye. Neymark. Investigation of the
Possible Application of Synthetic Zeolites as Carriers
and Catalysts for the Dehydrogenation and Cracking of
n-Paraffins 260
- Palek, M., P. Iru, O. Grubner, and G. Beyer.
Synthetic Zeolites as Molecular Sieves With Color
Indication of Water-Vapor Pressure 263
- Malyusov, V. A., N. N. Umnik, N. N. Kulov, N. M. Zhavoronkov,
G. I. Faydel', and D. O. Zisman. Purifying Formaldehyde
From Moisture and Formic Acid With the Aid of Synthetic
Zeolites 267

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GUTYRYA, V.S., glav. red.; KLIMENKO, A.P., zam. glav. red.; GALICH,
P.N., red.; KAMAKIN, N.M., red.; MAN'KOVSKAYA, N.K., red.;
MASUMYAN, V.Ya., red.; SERDYUK, O.P., red.

[Petroleum chemistry; paraffin petroleum hydrocarbons]
Neftekhimiia; parafinovye uglevodorody nefti, ikh vydelenie
i pererabotka. Kiev, Naukova dumka, 1964. 138 p.

(MIRA 17:10)

l. Akademiya nauk URSR, Kiev. Institut khimii vysokomole-
kulyarnykh soyedineniy.

FROLOVA, V.S.; POLATAYKO, R.I.; SKARCHENKO, V.K.; MUSIYENKO, V.P.;
GALICH, P.N.

Dehydrogenation of n-hexane on chrome-zinc oxide catalysts.
Neft. i gaz. prom. no.3:54-55 Jl-S '64.

(MIRA 17:12)

KONOVAL'CHIKOV, O.D.; GALICH, P.N.; MUSIYENKO, V.P.; SKARCHEVKO, V.K.;
PETROV, A.A.

Effect of the porous structure of an alumina-chromium oxide
catalyst on the conversion of n-hexane. Kin. i kat. 5 no.2:
250-354 Mr-Ap '64. (MIRA 17;8)

1. Institut khimii vysokomolekulyarnykh soyedineniy AN UkrSSR.

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genation. Kin. i kat. 5 no.3:548-549 My-Je '64.

(MIRA 17:11)

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SKARCHENKO, V.K.; FROLOVA, V.S.; GOLUBCHENKO, I.T.; MUSIYENKO, V.P.;
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1. Institut khimii vysokomolekulyarnykh soyedineniy AN UkrSSR.

MUSIYENKO, V.P.; BOLATAYEV, R.I.; SKARCHEVSKY, V.K.; FRONINA, V.G.;
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Ukr. khim. zhur. 30 no.9:915-918 '64.

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(MIRA 18:6)

BILAY, V.I.; PIDOPLICHKO, N.N. [Pidoplichko, M.M.]; GUTYRYA, V.S. [Hutyria, V.S.];
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KOVAL', E.Z.; MASUMYAN, V.Ya.; MIL'KO, A.A. [Mil'ko, O.O.]

Petroleum hydrocarbons as a source of carbon for microscopic
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GALICH, P.N.; GOLUBCHENKO, I.T.; GUTYRYA, V.S.; IL'IN, V.G.; NEYMARE, I.Ye.

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AN SSSR 161 no.3:627-628 Mr '65. (MIRA 18:4)

1. Institut khimii vysokomolekulyarnykh soyedineniy AN UkrSSR
i Institut fizicheskoy khimii im. Pisarzhevskogo AN UkrSSR.
2. Chlen-korrespondent AN SSSR (for Gutyrya).